

# The Chain Rule

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# The Chain Rule (case 1)

## Definition

- Suppose that  $z = f(x, y)$  is a differentiable function of  $x$  and  $y$ , where  $x = g(t)$  and  $y = h(t)$  are both differentiable functions of  $t$ . Then  $z$  is a differentiable function of  $t$  and

$$\frac{dz}{dt} = \frac{\partial f}{\partial x} \frac{dx}{dt} + \frac{\partial f}{\partial y} \frac{dy}{dt}.$$

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- If  $z = x^2y + xy^3$ , where  $x = \cos t$ ,  $y = \sin t$ , find  $dz/dx$  when  $t = \pi/2$ .

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- Find  $dz/dt$  if  $z = \sqrt{x^2 + y^2}$  and  $x = e^{2t}$  and  $y = e^{-2t}$ .

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- Find  $dz/dt$  if  $z = \sqrt{x^2 + y^2}$  and  $x = e^{2t}$  and  $y = e^{-2t}$ .
- The pressure  $P$  (in kilopascals), volume  $V$  (in liters), and temperature  $T$  (in kelvins) of a mole of an ideal gas are related by the equation  $PV = 8.31T$ . Find the rate at which the pressure is changing when the temperature is  $300K$  and increasing at a rate of  $0.1K/s$  and the volume is  $100 L$  and increasing at a rate of  $0.2 L/s$ .

## The Chain Rule (Case 2)

### Definition

- Suppose that  $z = f(x, y)$  is a differentiable function of  $x$  and  $y$ , where  $x = g(s, t)$  and  $y = h(s, t)$  are differentiable functions of  $s$  and  $t$ . Then

$$\begin{aligned}\frac{\partial z}{\partial s} &= \frac{\partial z}{\partial x} \frac{\partial x}{\partial s} + \frac{\partial z}{\partial y} \frac{\partial y}{\partial s} \\ \frac{\partial z}{\partial t} &= \frac{\partial z}{\partial x} \frac{\partial x}{\partial t} + \frac{\partial z}{\partial y} \frac{\partial y}{\partial t}\end{aligned}$$

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- $z = \ln(x^2 + y^2)$ , where  $x = e^s \cos t$  and  $y = e^s \sin t$ .
- $w = xy + xz + yz$ , where  $x = st$ ,  $y = e^{st}$ ,  $z = x + t$ .